CONCEPT: INTRODUCTION TO ENERGY

• Energy: the ability to perform _____.

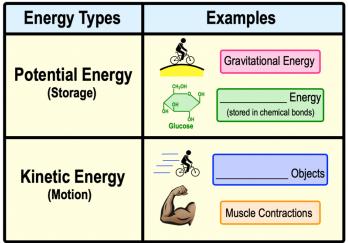
□ In Biology, **work** is a *transfer of energy* that causes an overall _____.

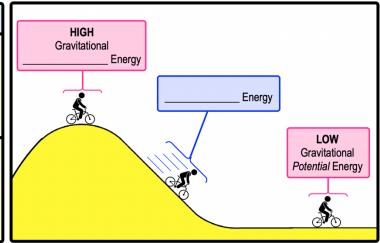
Potential vs. Kinetic Energy

1) _____ Energy: _____ energy that is available to do work.

2) ______ **Energy**: energy of ______.

EXAMPLE: Potential vs. Kinetic Energy.





PRACTICE: Which of the following is an example of potential energy?

- a) Water rushing over a water fall.
- b) A glucose molecule.
- c) A spring being released and expanding rapidly.
- d) An ant foraging (moving around) for food.

PRACTICE: Which of the following is NOT a form of kinetic energy?

- a) When the temperature of a substance rises causing the molecules of that substance to move faster.
- b) When a hockey puck slides across the ice after being hit by a player.
- c) When glucose molecules are broken down to create energy for cellular work.
- d) When the molecules of the air vibrate transmitting soundwaves.

CONCEPT: INTRODUCTION TO ENERGY

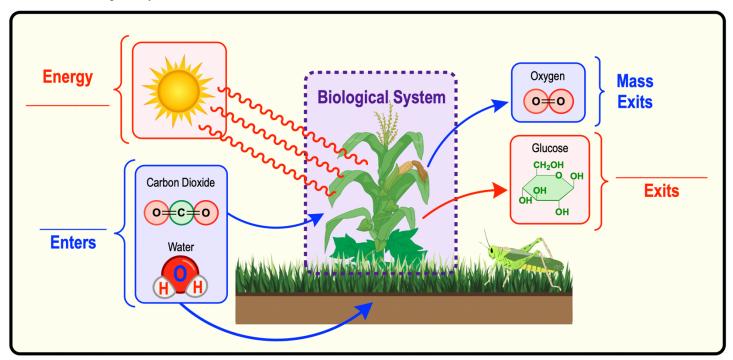
Thermodynamics: System vs. Surroundings

_____: the specific portion of matter being studied.

□ _____: everything else outside of the system.

□ Biological Systems transfer both _____ & ____ with the surroundings.

EXAMPLE: Biological Systems.



PRACTICE: Why are the principles of thermodynamics important for the study of biology?

- a) The principles of thermodynamics govern the chemical processes in all biological organisms.
- b) The principles of thermodynamics govern how an organism exchanges energy with their environment.
- c) The principles of thermodynamics govern how all life on Earth exchanges energy.
- d) All of the above.

PRACTICE: Which of the following is NOT an example of a biological system?

- a) A human eating a cheeseburger.
- b) A coral reef where energy and mass both enter and exit.
- c) The digestive system of an organism absorbing food energy and discarding food waste.
- d) The universe where the amount of energy and mass is constant.