

## CONCEPT: CATALYSIS

● Recall: Enzymes can lower \_\_\_\_\_ in several ways but the common theme is to \_\_\_\_\_ the *transition state*.

□ Question: What are the different types of enzyme catalysis mechanisms?

□ Enzyme catalysis is primarily conducted via \_\_\_\_\_ mechanisms:

1) \_\_\_\_\_ - \_\_\_\_\_ catalysis.    2) \_\_\_\_\_ catalysis.    3) \_\_\_\_\_ ion catalysis.    4) \_\_\_\_\_ catalysis.

**PRACTICE:** What is the common strategy for which enzyme catalysis occurs?

a) Increasing the probability of product formation.

c) Stabilization of the transition state.

b) Shifting the reaction equilibrium.

d) a and c are correct.

### 1) General Acid-Base Catalysis

● Acid-Base Catalysis: when an \_\_\_\_\_ or *base* catalyzes a reaction via a \_\_\_\_\_ ( $H^+$ ) transfer.

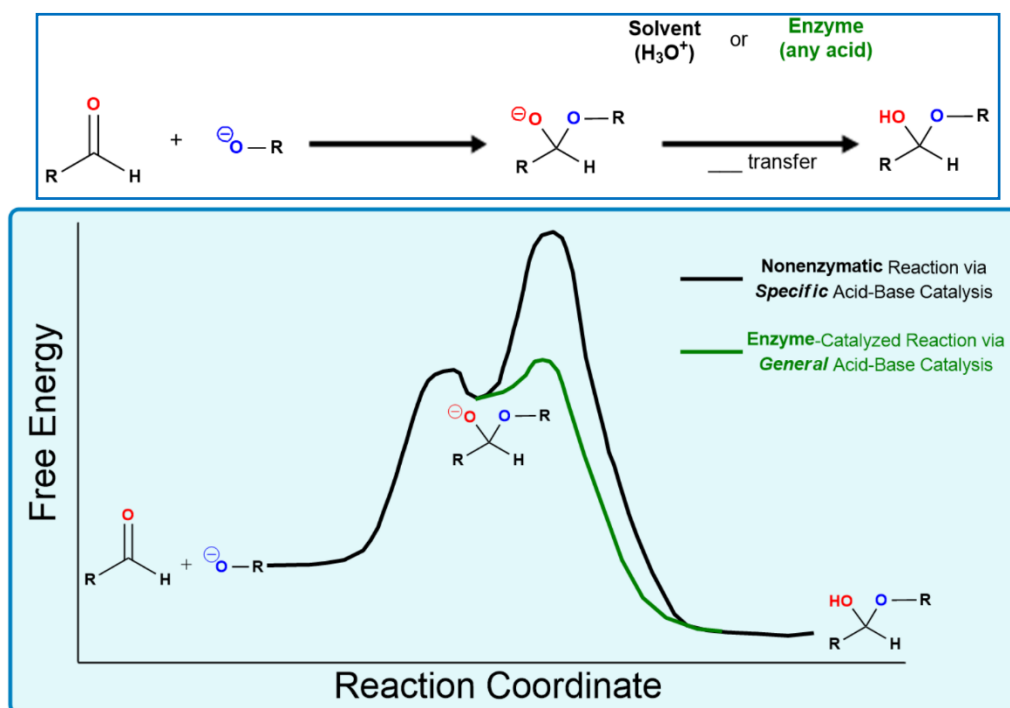
□ Unstable charged \_\_\_\_\_ can be stabilized with  $H^+$  transfers.

● \_\_\_\_\_ main types of acid-base catalysis:

□ *Specific* Acid-Base Catalysis: *only* solvent ( $H_2O$ ) serves as a  $H^+$  transfer source; sometimes this is too \_\_\_\_\_.

□ \_\_\_\_\_ *Acid-Base Catalysis*: the \_\_\_\_\_ active site *mediates*  $H^+$  transfers via *any* acid/base.

**EXAMPLE:** Specific vs. General Acid-Base Catalysis.



**PRACTICE:** Which of the following could not be the direct  $H^+$  transfer source for general acid-base catalysis?

a) Lysine.

b) Glutamic acid.

c) Water.

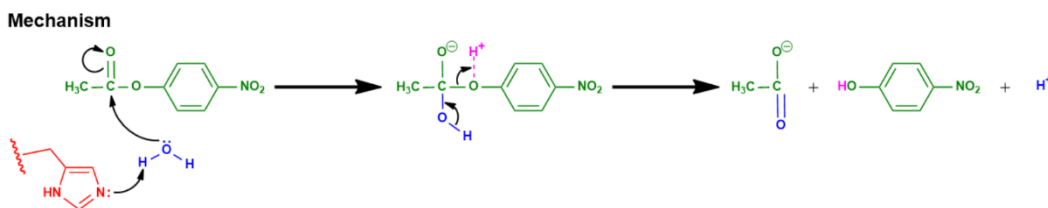
d) Tyrosine.

e) Phenylalanine.

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**PRACTICE:** The catalytic mechanism below is an example of:

- a) General acid catalysis.
- b) Specific acid catalysis.
- c) General base catalysis.
- d) Specific base catalysis.



**PRACTICE:** Which of the following best applies to general acid-base catalysis?

- a) A proton is transferred between the enzyme and substrate.
- b) Uses nucleophilic functional groups.
- c) May take part in interactions involving Fe<sup>2+</sup>.
- d) Catalyst retains its original form after reaction occurs.
- e) Other than losing/gaining a H<sup>+</sup>, the catalyst retains its original form.
- f) a & d.
- g) a & e.