## **CONCEPT:** LIPIDS

- Fatty acids have a unique naming convention
  - $\Box$  # of carbons : # of double bonds ( $\Delta$ # of carbons participating in double bonds)
  - □ Omega 3 and 6 fatty acids must be acquired through diet because humans can't synthesize them

- Naturally occurring unsaturated fatty acids are always cis
  - □ Melting point increases as chain length increases
    - The longer the chain, the less soluble the fatty acid
  - □ Unsaturation greatly reduces melting point of the fatty acids
    - Trans fats have higher melting points because, despite unsaturation, their trans double bonds are linear
- Triacylglycerols storage lipids made from glycerol and 3 fatty acids attached via ester bonds
  - □ If one unsaturated chain, always on carbon 2