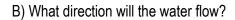
CONCEPT: OSMOSIS

CONCEPT: OSIVIOSIS
• Diffusion: substance movement from high concentrations to low concentrations of the same substance.
● <u>Osmosis</u> : diffusion of a solvent (usually) across a <i>semi-permeable</i> membrane.
□ <u>Osmotic pressure</u> : pressure required to prevent the flow of solvent.
Osmosis direction depends on the (or the relative concentration of solutes dissolved in the solutions).
□ solutions have the <i>same solute</i> concentration.
□ solutions have <i>lower <u>solute</u></i> concentrations.
□ solutions have <i>higher</i> <u>solute</u> concentrations.
EXAMPLE: Label the tonicity of the outside solution.
Outside Outside Outside Inside
Osmosis Direction
●H₂O will move from to solutions if solutes cannot diffuse across the membrane.
□ Water moves toward the more concentrated solute solution to dilute it until it is
 Water still moves from higher concentrations of <u>water</u> to lower concentrations of <u>water</u>.
□ <u>Hypo</u> tonic solutions: lower solute concentrations but H₂O concentration.
□ <u>Hyper</u> tonic solutions: higher solute concentration but H₂O concentration.
EXAMPLE: Direction of Osmosis HYPO Lower [solute] Higher [H ₂ O]
Results of Osmosis
●Hyp_tonic environments: cause cells to swell like a hipp & potentially lyse (rupture/burst).
□ Cells with do <i>not</i> lyse in hypotonic solutions (membrane expansion prevented).
□ Preferred by <i>plant</i> cells due to increased (water pressure on cell membrane).
●Hyptonic environments: dehydrate cells like a hyper-kid gets dehydrated. Hyper-kids get dehydrated.
Environment: Hypotonic Isotonic Hypertonic Environment: Hypotonic Isotonic Hypertonic
Animal Cells Plant Cells

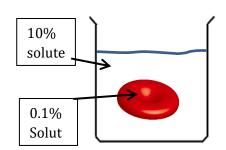
CONCEPT: OSMOSIS

PRACTICE: A) What is the tonicity of the outside solution in comparison to the cell?

- a) Hypotonic
- b) Isotonic
- c) Hypertonic
- d) electrotonic



- a) Inside → Outside
- b) Outside → Inside
- c) Water flows in both directions.
- d) No flow of water.



PRACTICE: Plants become turgid when placed in this type of solution:

- a) Hypotonic
- b) Isotonic
- c) Hypertonic
- d) Megatonic