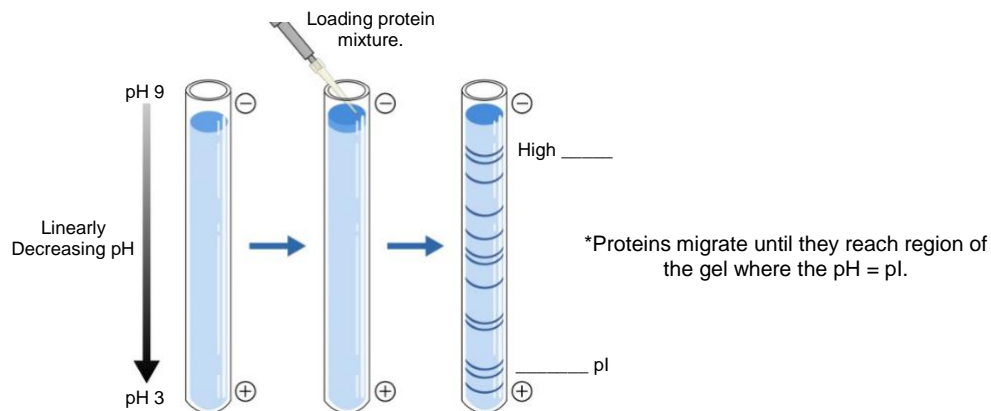


## CONCEPT: ISOELECTRIC FOCUSING

- **Isoelectric Focusing (IEF):** electrophoresis technique separating proteins only based on \_\_\_\_\_ points (pI).
  - Recall: pI is the pH where the net charge of the protein is \_\_\_\_\_.
- A stable/immobile \_\_\_\_\_ gradient is established into the gel.
  - Proteins alter their \_\_\_\_\_ as they migrate through different regions of the gel with different pH.
  - Proteins continue to migrate until they reach the portion of the gel that has a pH \_\_\_\_\_ to its pI.
  - When the pH = pI of a protein, it has a *neutral* net charge of zero & does \_\_\_\_\_ migrate in an electric field.

**EXAMPLE:** Isoelectric focusing (IEF).



**PRACTICE:** At some point during isoelectric focusing, proteins stop moving through the gel because:

- The proteins do not have ionized groups at that pH.
- The proteins have a net charge of zero at that pH.
- The proteins have a net positive or net negative charge at that pH.
- Their mass is too large to be moved at that position in the gel.

**PRACTICE:** Electrophoretic separation at pH 6 of a sample mixture with Peptide #1 (MW 100) Peptide #2 (MW 200) and Peptide #3 (MW 400) would result in which of the following? (Note: the pI of each peptide occurs at pH 6).

- Peptide #1 would move the farthest.
- Peptide #2 would move the farthest.
- Peptide #3 would move the farthest.
- None of the peptides would move.

**PRACTICE:** Mark the approximate final position of the following tripeptide on the isoelectric focusing gel: Glu-Met-Asp.

Hint: calculate the isoelectric point of the peptide.

Ionizable Group		pK <sub>a</sub>
C-terminus	—COOH	3.5
Asp	—C(H <sub>2</sub> )—COOH	3.9
Glu	—C(H <sub>2</sub> )—C(H <sub>2</sub> )—COOH	4.1
N-terminus	—NH <sub>3</sub> <sup>+</sup>	8.0

