

CONCEPT: ISOELECTRIC POINT OF AMINO ACIDS WITH IONIZABLE R-GROUPS

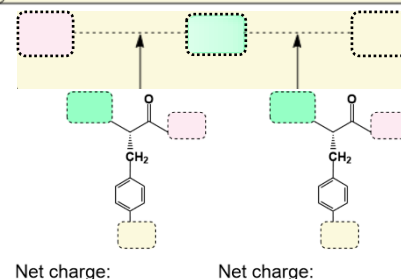
● Amino acids with ionizable R-groups have ____ pK_a values, but pI is still an average of only ____ pK_a values.

□ Question: So how do you know which pK_a values to average?

- 1) Know the net charge of the protonated & deprotonated forms of ____ R-groups.
- 2) Order the 3 pK_a values from ____ to largest & use the middle pK_a as a guide.
- 3) Determine *net* ____ of predominate amino acid structures at any pH between adjacent pK_a values.
- 4) Average ____ pK_a s: the middle pK_a with the pK_a closest to the pH that gives a neutral net charge.

EXAMPLE: Calculate the isoelectric point of Tyr. $pK_{a1} = 9.1$. $pK_{a2} = 2.2$. $pK_R = 10.1$.

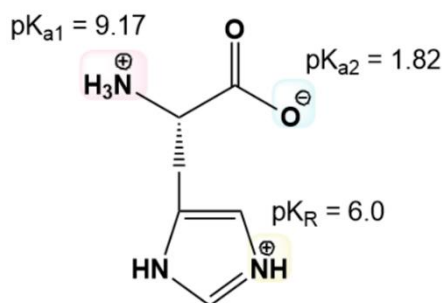
Step #1: ____ how Amino Acid R-groups ionize.	Step #2: ____ the 3 pK_a values from smallest to greatest. $pK_{a1} < pK_{a2} < pK_{a3}$	Step #3: Determine net ____ of predominate structures at any pH between each pair of adjacent pK_a 's (by comparing pH to all pK_a values). pH: pK_{a1} pK_{a2} pK_{a3}
Step #4: Calculate pI by ____ middle pK_{a2} & the pK_a "sandwiching" the pH where the predominate structure has a <i>neutral</i> net charge. $pI = \frac{pK_{a2} + (\text{either } pK_{a1} \text{ or } pK_{a3})}{2}$		



- a) 7.48
- b) 3.22
- c) 6.96
- d) 5.65

PRACTICE: What is the pI of His?

- a) 7.59
- b) 6.58
- c) 3.91
- d) 5.5



PRACTICE: Electrophoresis separates molecules by migration through a gel only if they have a net charge. How do you expect an amino acid to migrate during electrophoresis when the $pH = pI$?

- a) Migrate from the (-) end to the (+) end.
- b) Migrate from the (+) end to the (-) end.
- c) No migration.

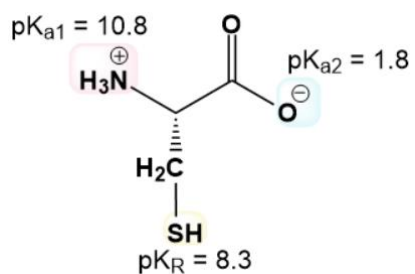
CONCEPT: ISOELECTRIC POINT OF AMINO ACIDS WITH IONIZABLE R-GROUPS

PRACTICE: Draw Glu and calculate its isoelectric point. $pK_{a1} = 9.67$. $pK_{a2} = 2.19$. $pK_R = 4.25$.

- a) 7.48
- b) 3.22
- c) 6.96
- d) 5.93

PRACTICE: Calculate Cys's pl.

- a) 2.3
- b) 3.14
- c) 5.05
- d) 6.05



PRACTICE: Calculate the pl of Arg. $pK_{a1} = 9$. $pK_{a2} = 2.2$. $pK_R = 12.5$.

- a) 1.9
- b) 10.75
- c) 3.6
- d) 12.5

PRACTICE: Calculate the pl of Asp.

- a) 1.4
- b) 3
- c) 5.95
- d) 6.85

