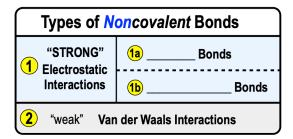
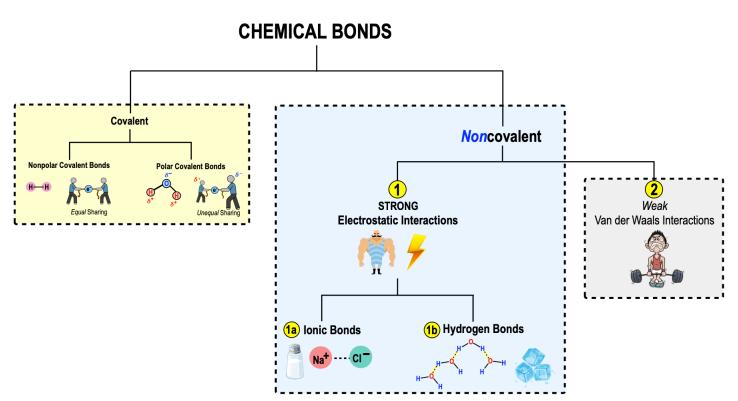
CONCEPT: NONCOVALENT BONDS

- •_____-covalent Bonds: interactions between 2 atoms resulting from full or partial ______.
 - □ Unlike covalent bonds, there is _____ sharing of electrons in noncovalent bonds.
 - □ Several types of *noncovalent bonds* are commonly found in biology:





PRACTICE: Which of the following are considered to be very weak non-covalent chemical bonds?

- a) Polar covalent bonds.
- b) lonic bonds.
- c) Non-polar covalent bonds.
- d) Van der Waals bonds.
- e) Hydrogen bonds.
- f) Electrostatic bonds.