## **CONCEPT: EXPERIMENTAL ERROR**

When any calculation is done there is a level of error involved.

- Error can be grouped into 2 categories: random (lack of \_\_\_\_\_\_) and systematic (lack of \_\_\_\_\_\_).
- Random errors are unpredictable and can lead to results that are either too high or too low.
- Systematic errors are more predictable and can lead to results that are always too high or always too low, but not both.
- In most cases, a percent error of less than \_\_\_\_\_ will be acceptable.

% error = 
$$\frac{|\text{Experimental value} - \text{Theoretical value}|}{\text{Theoretical value}} \times 100$$

The percent error formula is a useful tool for determining the \_\_\_\_\_\_ of your calculations.

• The \_\_\_\_\_ value is your calculated value, and the \_\_\_\_\_ value is

**EXAMPLE 1:** Which of the following features are indicative of random errors?

- i. Doing numerous measurements and taking the average in order to minimize any errors.
- ii. The results of an experiment are consistently greater than expected or less than expected.
- iii. Refining the parameters of the experiment help to eliminate any errors.
- iv. The exist of the error is hard to determine.

your known value.

**EXAMPLE 2:** Which of the following represent a systematic error when measuring the mass of an anhydrous object?

- i. You weigh the object before all the water has evaporated.
- ii. The scale used has not been properly calibrated.
- iii. Airflow near the balance causes the precise mass to vary.
- iv. You write down the incorrect mass of the anhydrous object.