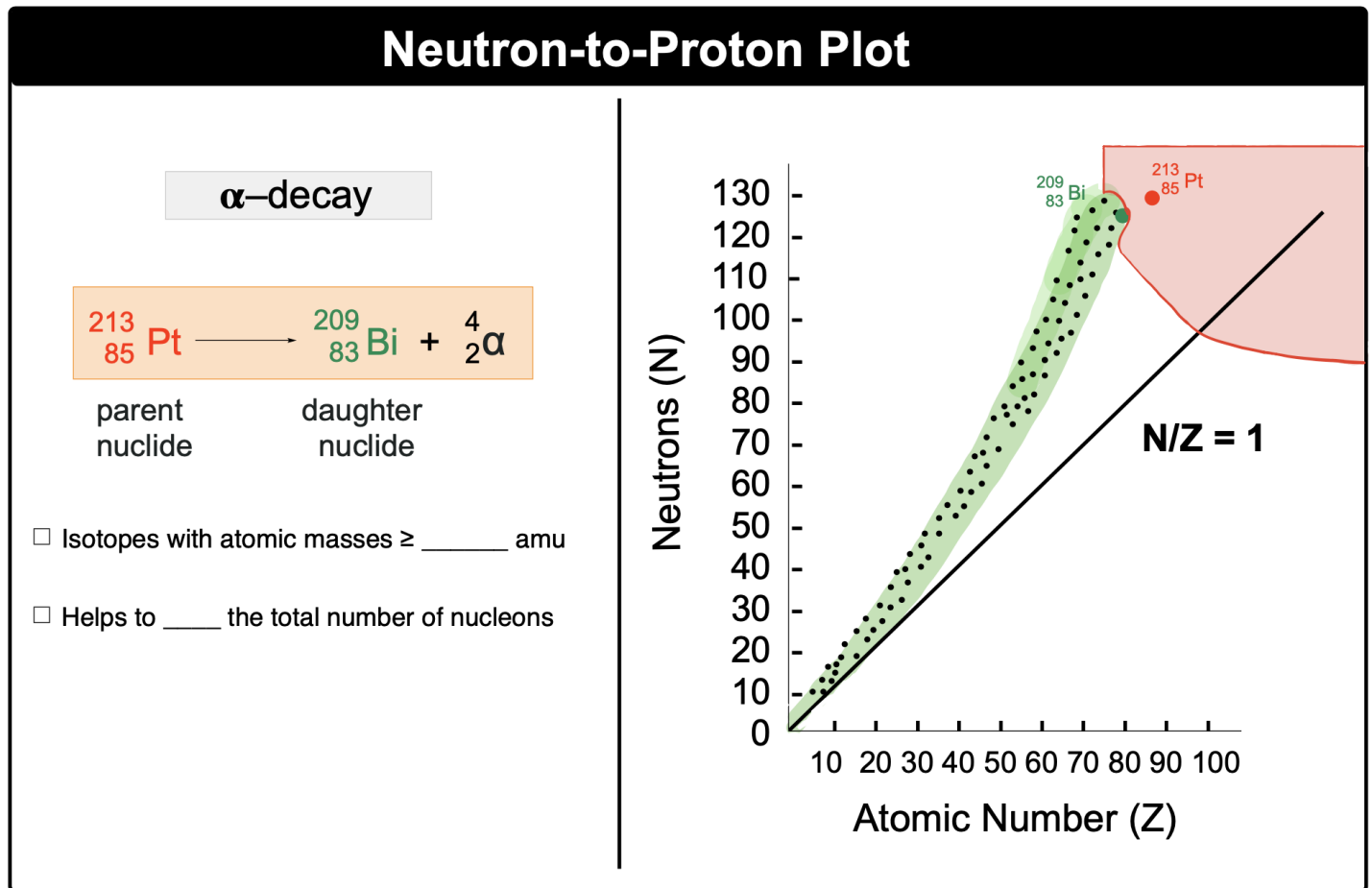


## CONCEPT: BAND OF STABILITY: ALPHA DECAY & NUCLEAR FISSION

- Isotopes that lie outside the Band (Valley) of Stability are considered \_\_\_\_\_, \_\_\_\_\_ isotopes.
  - These isotopes will alter their number of \_\_\_\_\_ and/or \_\_\_\_\_ to move closer to the Band (Valley) of Stability.
  - They do this predominantly by alpha decay, nuclear fission, \_\_\_\_\_ decay, \_\_\_\_\_ capture, or \_\_\_\_\_ emission.

### Alpha Decay

- Alpha Decay** happens for isotopes in the \_\_\_\_\_ corner of the N/Z plot.
  - These isotopes have an \_\_\_\_\_ of neutrons and protons.



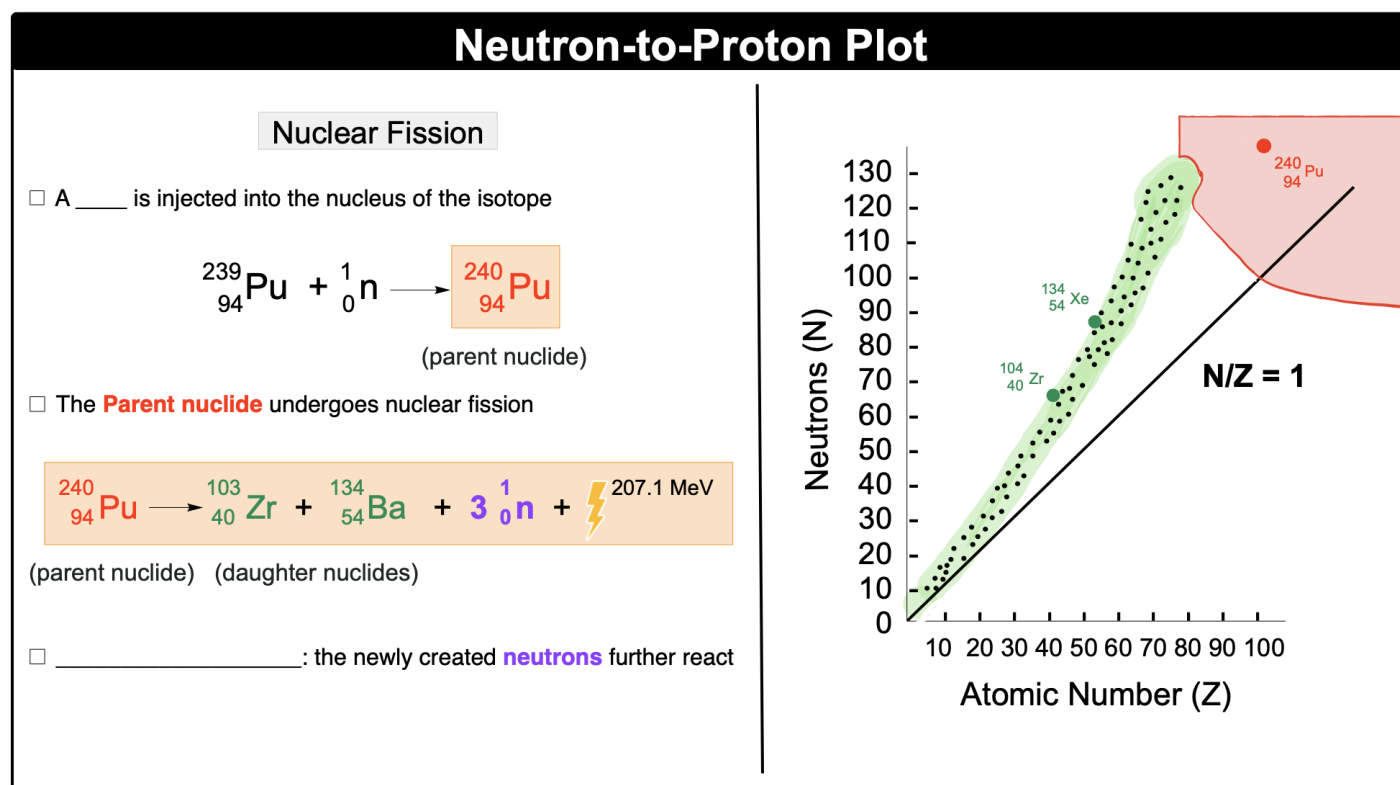
**EXAMPLE:** Which daughter nuclide would reside in the band of stability created from the alpha decay of lead-212?

- a) Polonium-216                      b) Mercury-208                      c) Thallium-212                      d) Lead-209

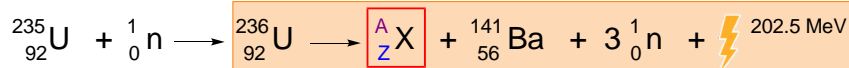
## CONCEPT: BAND OF STABILITY: ALPHA DECAY & NUCLEAR FISSION

### Nuclear Fission

- Under **Nuclear Fission** a(n) \_\_\_\_\_ is shot at the nucleus of an isotope and \_\_\_\_\_ an extremely large amount of energy.
  - **Additional Benefit:** large heavy elements (greater than \_\_\_\_\_ amu) are split into 2 lighter daughter nuclides.
  - Drastically \_\_\_\_\_ the total number of nucleons for an isotope.



**EXAMPLE:** Nuclear fission is a commonly occurring process for uranium-235. Provide the identity of the missing daughter nuclide produced at the end of the reaction.



a) Cesium-77

b) Krypton-92

c) Bromine-81

d) Krypton-94

**PRACTICE:** Which of the following is a potential daughter nuclide created from the nuclear fission of uranium-233 that resides near the band of stability?

a) Strontium-94

b) Radon-222

c) Curium-247

d) Thorium-232