## **CONCEPT: ALDEHYDE AND KETONE REACTIONS**

- The reducing agent of \_\_\_\_\_ dissolved in water reacts with an alcohol.
  - □ **Reducing Agent:** the compound used to \_\_\_\_\_ aldehydes and ketones.
    - The carbonyl oxygen gains an \_\_\_\_\_ and the carbonyl carbon gains an \_\_\_\_\_.

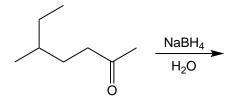
**EXAMPLE**: Determine the alcohol product formed in the following reaction.

$$\begin{array}{c} O \\ \parallel \\ CH_3CH_2CCH_2CH_3 \end{array} \xrightarrow{\begin{array}{c} NaBH_4 \\ \hline H_2O \end{array}}$$

**PRACTICE:** Determine the alcohol product formed in the following reaction.

## **CONCEPT: ALDEHYDE AND KETONE REACTIONS**

**PRACTICE:** Name the alcohol product formed from the following reduction reaction.



- a) 5-ethyl-2-pentanol
- b) 3-methyl-6-heptanol c) 5-methyl-2-heptanol
- d) 3-methyl-5-pentanal

PRACTICE: Which of the following compounds could not be reduced?

- a) 2,2-dimethylpentane
- b) 2-methyl-1-pentanal
- c) 3-ethyl-2-heptanone
- d) 4-bromoheptanoic acid