• Recall: Carboxylic acids possess a carbonyl carbon connected to a \_\_\_\_\_\_(OH) group.

• Set of rules for naming carboxylic acids are similar to aldehydes.

 $\hfill\Box$  The carbonyl carbon of the carboxylic acid is always numbered  $\hfill$ 

□ Modify the ending from -\_\_\_\_ to -\_\_\_\_.

location-substituent-parent-modifier

**EXAMPLE**: Provide the systematic name for the following carboxylic acid.

$$\begin{array}{c} \operatorname{Br} & \operatorname{O} \\ | & \| \\ \operatorname{CH}_3\operatorname{CH}_2\operatorname{CH}\operatorname{CH}\operatorname{CH}_2\operatorname{C}\operatorname{-OH} \\ | & \\ \operatorname{CH}_2\operatorname{CH}_3 \end{array}$$

**STEP 1:** Find the longest carbon chain (parent chain) and assign name according to the prefixes and modifier.

□ Parent chain should include the COOH group and have \_\_\_\_\_ number of carbons.

 $\hfill\Box$  If a tie between longest chains, choose chain with more substituents.

**STEP 2:** Assign name to all the substituents.

**STEP 3:** Start numbering the chain at the carbon of the \_\_\_\_\_ group.

STEP 4 to 6: Repeat steps from previous naming topics.

**PRACTICE:** If the substituent name of the OH group is hydroxy in the presence of a carboxylic acid, provide the systematic name for the following compound.

### **Common Naming**

- Simple carboxylic acid name consists of a common name prefix and ends with -\_\_\_\_\_.
- Substituted carboxylic acids utilize Greek symbols ( \_\_\_\_\_\_\_) to indicate location of substituents.
  - □ Greek symbols are assigned to carbons adjacent to the \_\_\_\_\_ carbon.

$$\begin{pmatrix}
H_2 & \parallel \\
H_3C & C & C \\
C & C & C
\\
H_2 & H_2
\end{pmatrix}$$
OH

**EXAMPLE**: Provide a common name for the following carboxylic acid.

- **STEP 1:** Find the longest carbon chain and assign name according to common name prefixes and ending.
  - □ Parent chain should include the COOH group and have \_\_\_\_\_ number of carbons.
  - □ If a tie between longest chains, choose chain with more substituents.
- **STEP 2:** Assign name to all the substituents.
- STEP 3: Assign Greek (location) for each substituent.
  - □ When more than 1 identical substituents, use prefixes: \_\_\_\_\_ 2, \_\_\_\_ 3, \_\_\_\_\_ 4.
- STEP 4: Name substituents in \_\_\_\_\_ order; prefixes do not count.
- STEP 5: Use \_\_\_\_\_\_ between Greek symbols, and \_\_\_\_\_ to separate Greek symbols from letters.
  - □ Letters are not separated from letters.

**PRACTICE:** Provide a common name for following molecule.

## Naming Dicarboxylic Acids

- Dicarboxylic acids contain 2 \_\_\_\_\_ groups.
- IUPAC: same rules as carboxylic acids but replace -\_\_\_\_ suffix with -\_\_\_\_.
  - □ Parent chain name is not modified.
- Common names are used more often for dicarboxylic acids.

MEMORY TOOL: Oh My Such Good Apple Pie.

НО			
ООН			
IUPAC: ethane acid			
Common: oxalic acid			

Dicarboxylic Acid Common Names					
Oxalic acid	НООССООН	Glutaric acid	HOOC(CH <sub>2</sub> )_COOH		
Malonic acid	HOOCCH₂COOH	Adipic acid	HOOC(CH <sub>2</sub> )_COOH		
Succinic acid	HOOC(CH <sub>2</sub> )_COOH	Pimelic acid	HOOC(CH <sub>2</sub> )_COOH		

**EXAMPLE**: Provide a common name for the following carboxylic acid.

**STEP 1:** Find the longest carbon chain and assign name according to common name prefixes.

 $\hfill\Box$  Parent chain should include both \_\_\_\_\_ groups.

STEP 2-5: Repeat from Common Naming of Carboxylic Acids.

**PRACTICE:** Give IUPAC name for the following dicarboxylic acid.

 $\textbf{PRACTICE:} \ \, \text{Draw a structure for the given common name: } \alpha\text{-bromo-}\beta\text{-hydroxyadipic acid.}$