

## CONCEPT: REDOX REACTIONS

● **Oxidation-Reduction** ( ) **Reaction:** *transfers* (e<sup>-</sup>) between molecules.


- **Oxidation:** the process of one or more electrons.
- **Reduction:** the process of one or more electrons (overall charge is reduced).
- Oxidation & reduction reactions always occur (at the same time).

**EXAMPLE:** Redox Reaction (**LEO** the Lion goes **GER**).


**LEO** the lion goes **GER**


lose electrons oxidation


gain electrons reduction




LEO



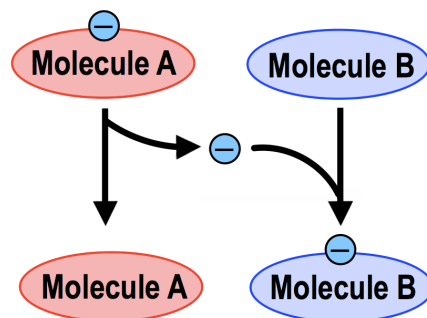






**EXAMPLE:** Which of the following molecules is oxidized and which is reduced?

- a) Molecule A is oxidized ; Molecule B is reduced.
- b) Molecule A is reduced ; Molecule B is oxidized.
- c) Both Molecule A & Molecule B are reduced.
- d) Both Molecule A & Molecule B are oxidized.



**PRACTICE:** Oxidation is the \_\_\_\_\_, and reduction is the \_\_\_\_\_.

- a) Gain of electrons ; Loss of electrons.
- b) Gain of protons ; Loss of protons.
- c) Loss of electrons ; Gain of electrons.
- d) Gain of oxygen ; Loss of oxygen.

**PRACTICE:** When glucose donates electrons to NAD<sup>+</sup> creating NADH, the glucose molecule becomes:

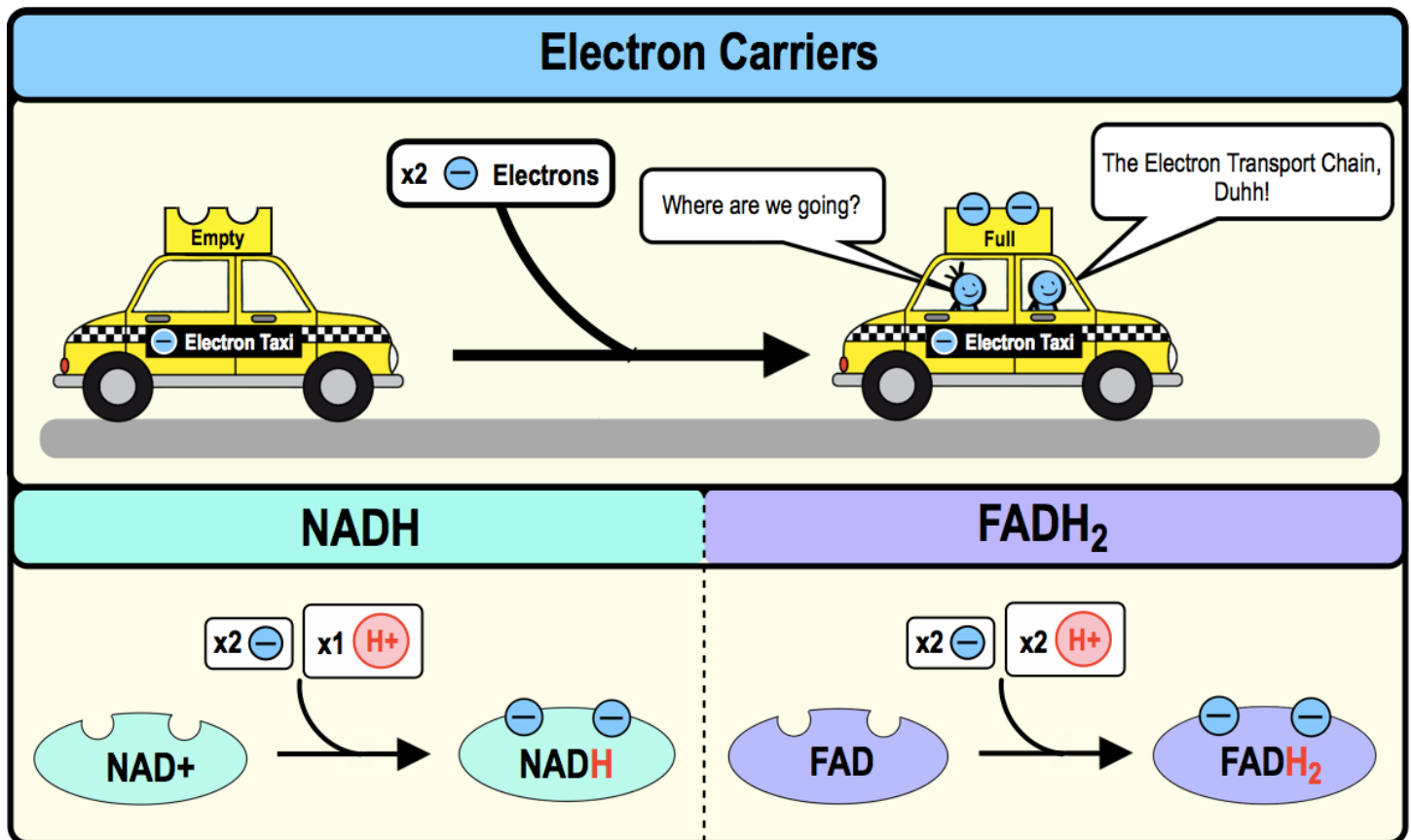
- a) Hydrolyzed.
- b) Oxidized.
- c) Neutral.
- d) Reduced.

## CONCEPT: REDOX REACTIONS

### Electron Carriers: NADH & FADH<sub>2</sub>

- Many biological processes (including *Cellular Respiration*) conduct redox reactions using electron \_\_\_\_\_.
- Electron carriers such as \_\_\_\_\_ & \_\_\_\_\_ each carry/transport \_\_\_\_\_ electrons.
- NADH & FADH<sub>2</sub> can shuttle electrons to different locations within a cell like an electron “\_\_\_\_\_” cab.
- NAD<sup>+</sup> & FAD are the \_\_\_\_\_ forms of NADH & FADH<sub>2</sub>, respectively.

**EXAMPLE:** Formation of NADH and FADH<sub>2</sub>.



- Helpful Memory Tool:* think that NAD\_\_\_\_\_ and FAD\_\_\_\_\_ are just a little bit “\_\_\_\_\_eavier.”
- During *Cellular Respiration*, electrons carried by NADH & FADH<sub>2</sub> go to the *Electron Transport Chain*.

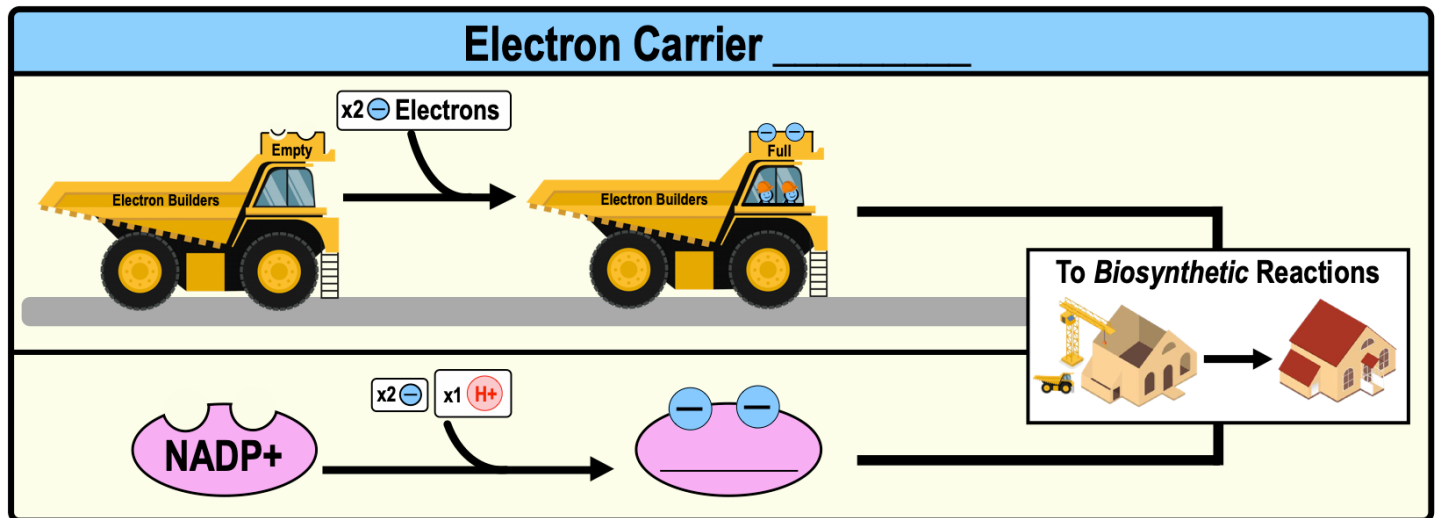
**PRACTICE:** An electron carrier before it harvests energy from glucose molecules in a series of gradual steps is:

- a) Pyruvate.                      b) AMP.                      c) ATP.                      d) NAD<sup>+</sup>.                      e) NADH.

## CONCEPT: REDOX REACTIONS

### Electron Carrier NADPH

- NADPH is an electron \_\_\_\_\_ similar to NADH and  $\text{FADH}_2$  for reduction reactions.
- However, unlike NADH &  $\text{FADH}_2$ , NADPH is used in \_\_\_\_\_ reactions for biosynthesis.



**PRACTICE:** NADH is commonly used as an electron carrier during the *breaking down* of complex molecules like glucose in cellular respiration. NADPH is also a common electron carrier. However, NADPH is used to *build* complex molecules like glucose in a process called:

- Biogenesis.
- Biocatalysis.
- Biosynthesis.
- Metabolism.

**PRACTICE:** Which electron carrier is *not* matched with its correct function.

- NADH: Shuttles electrons to regions of the cell which need energy.
- $\text{FADH}_2$ : Shuttles electrons to the electron transport chain during cellular respiration.
- NADPH: Shuttles electrons to photosynthesis where glucose molecules are being made.
- All of the above electron carriers are matched with their correct function.