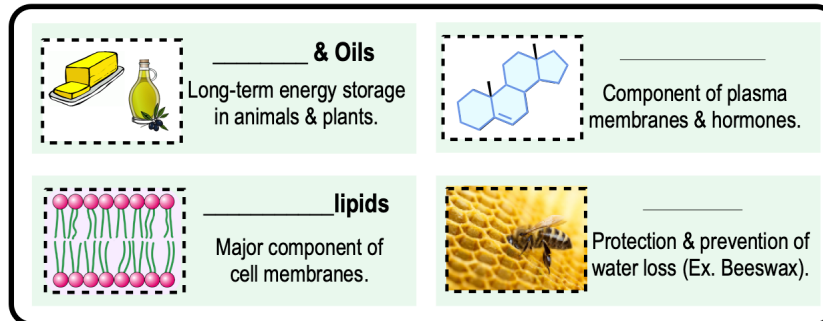


## CONCEPT: LIPIDS

- \_\_\_\_\_: *hydrophobic* biomolecules insoluble in water that are highly diverse in their *structure & function*.
  - Can also be \_\_\_\_\_: having \_\_\_\_\_ *hydrophobic & hydrophilic* groups.
  - Do \_\_\_\_\_ form polymers (*unlike* the other classes of biomolecules).
  - Lipids include: *fats & oils, phospholipids, steroids, & waxes*.

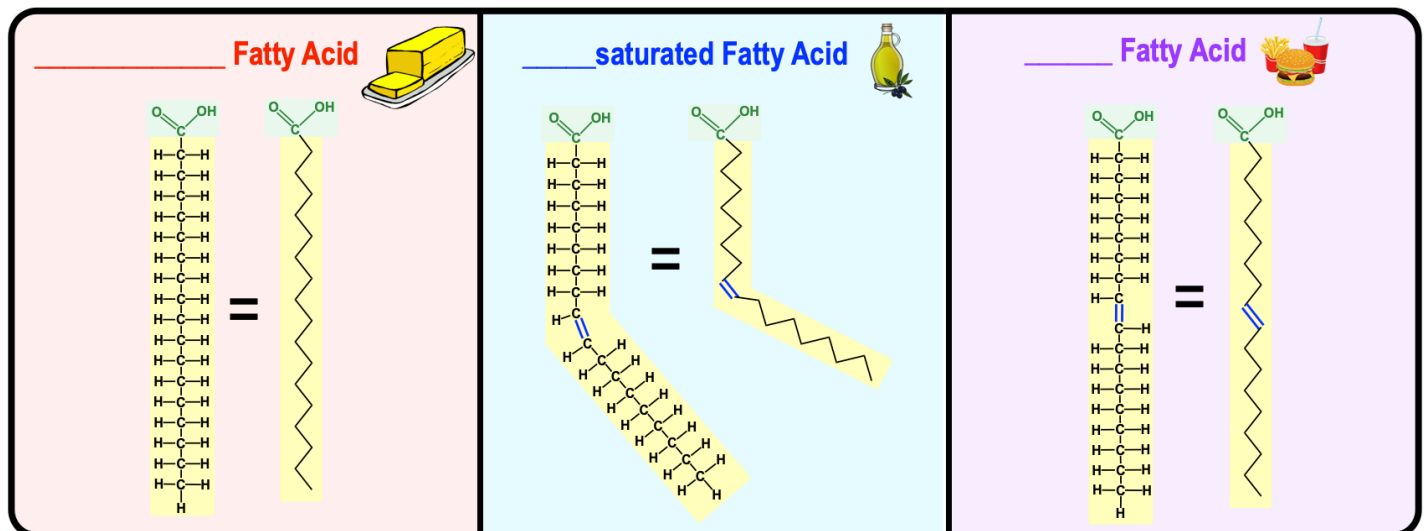
**EXAMPLE:** Types of Lipids.



## Fatty Acids

- **Fatty Acids:** *hydrocarbon* chains of varying length with a \_\_\_\_\_ *acid*.
  - 1) **Saturated Fatty Acids:** fully \_\_\_\_\_ with hydrogens (only contains C-C \_\_\_\_\_ bonds).
    - \_\_\_\_\_ at room temp.
  - 2) **Unsaturated Fatty Acids:** \_\_\_\_\_ fully saturated with hydrogens due to presence of  $\geq 1$  C=C \_\_\_\_\_ bond.
    - Double bond creates a bend or a “\_\_\_\_\_” in the chain, making them \_\_\_\_\_ at room temp.
    - \_\_\_\_\_ **Fats:** artificial *unsaturated fatty acids* that are NOT kinked (linear).

**EXAMPLE:** Saturated, Unsaturated & Trans Fats.



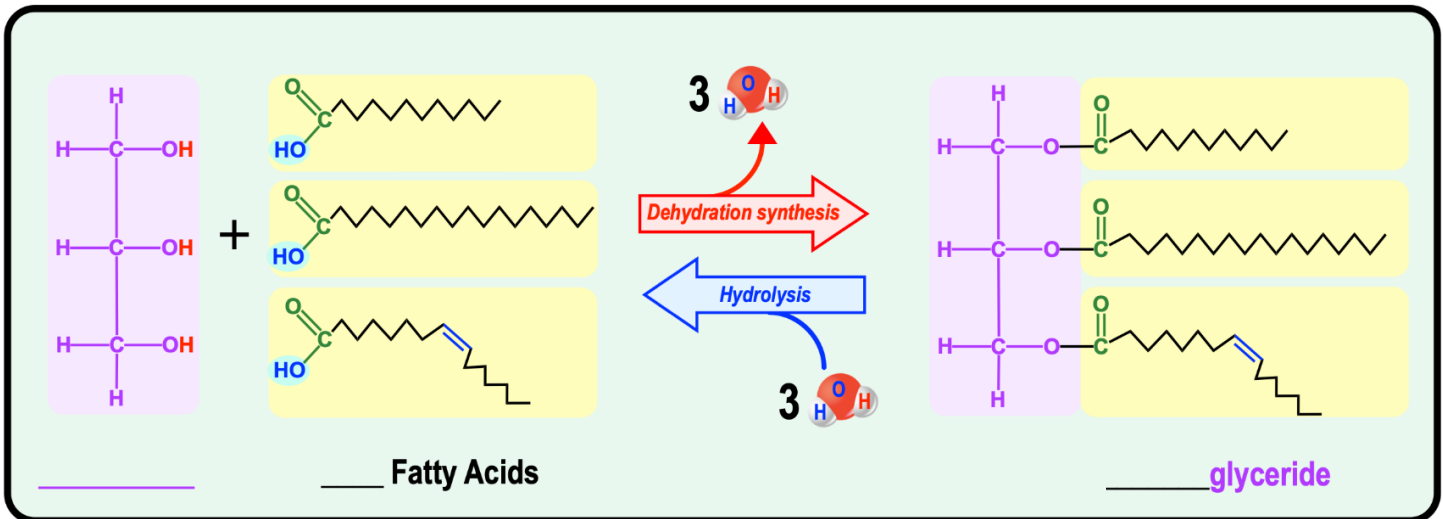
**PRACTICE:** Which type of fatty acids only contain Carbon-Carbon single bonds?

- a) Unsaturated.      b) Saturated.      c) Trans fats.      d) Steroids.

## CONCEPT: LIPIDS

### Triglycerides

- **glycerides:** a lipid with \_\_\_\_\_ fatty acid chains covalently linked to a single \_\_\_\_\_ molecule.
  - Fatty acids linked to glycerol via \_\_\_\_\_ synthesis reactions.

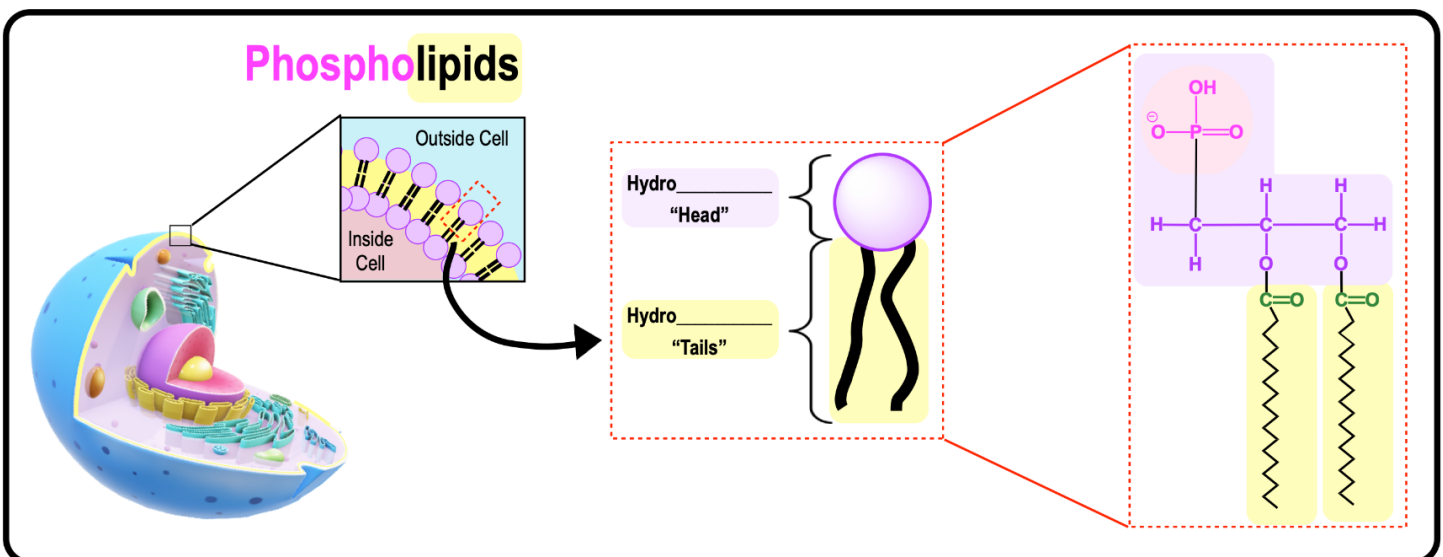


**PRACTICE:** A triglyceride is a form of \_\_\_\_\_ composed of \_\_\_\_\_.

- a) Lipid ; fatty acids & Glucose.
- b) Lipid ; Fatty acids & Glycerol.
- c) Carbohydrate ; Fatty acids only.
- d) Lipid ; Ribose.

### Phospholipids

- **Phospholipids:** large class of *lipids* that contain a \_\_\_\_\_ group.
  - Major component of ALL \_\_\_\_\_.
  - *Amphipathic* molecules with a \_\_\_\_\_ head and \_\_\_\_\_ tails.



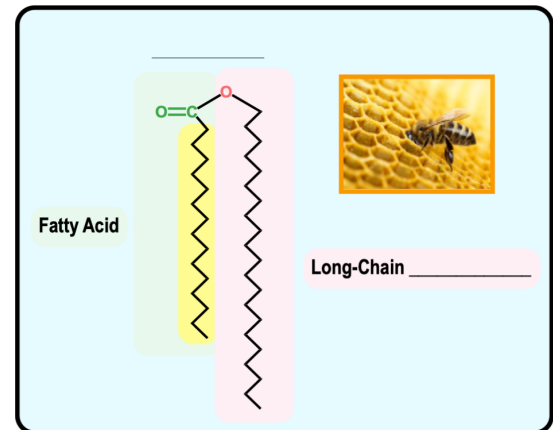
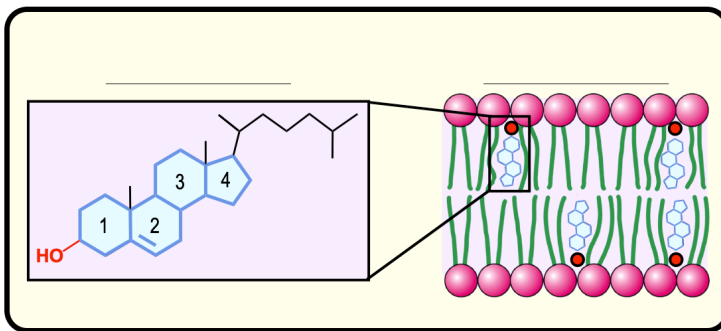
## CONCEPT: LIPIDS

**PRACTICE:** How do phospholipids interact with water molecules?

- a) The polar heads avoid water; the nonpolar tails attract water (because water is polar and opposites attract).
- b) Phospholipids do not interact with water because water is polar and lipids are nonpolar.
- c) The polar heads interact with water; the nonpolar tails do not.
- d) Phospholipids dissolve in water.

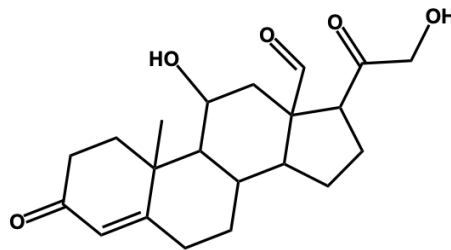
## Steroids & Waxes

- \_\_\_\_\_: lipids that are made of \_\_\_\_\_ fused carbon ring structures.
  - **Cholesterol:** common steroid important for the *structure* of animal cell \_\_\_\_\_.
- \_\_\_\_\_: another class of lipid that can be used for protection & prevention of water loss.



**PRACTICE:** The molecule shown the figure is a \_\_\_\_\_.

- a) Fatty acid.
- b) Wax.
- c) Steroid.
- d) Triacylglycerol.
- e) Phospholipid.



**PRACTICE:** Choose the correct statement about biological waxes:

- a) They contain at least one steroid molecule.
- b) They are fatty acids bound to long chain alcohol molecules.
- c) They are extremely hydrophilic.
- d) They are made of 4 fused carbon ring molecules.