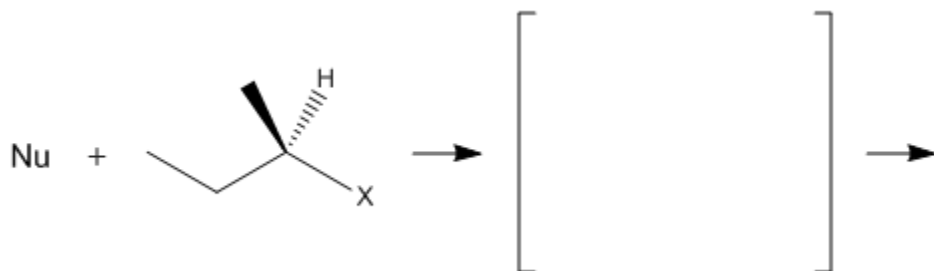
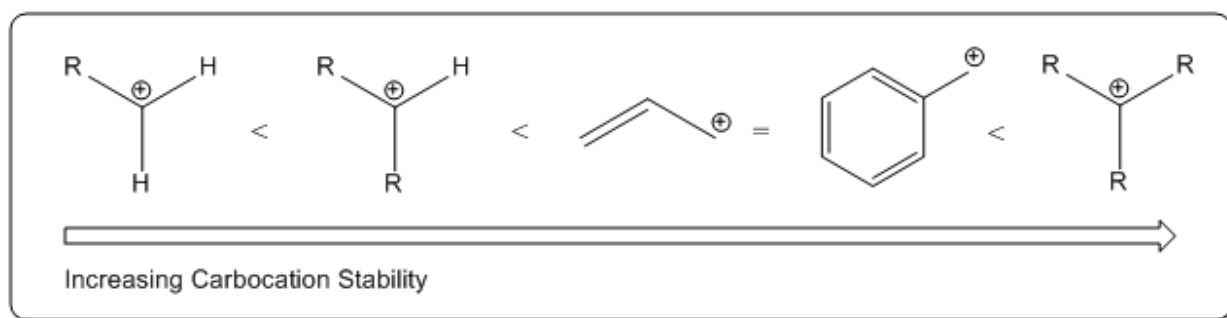


CONCEPT: S_N1 REACTION

□ A neutral nucleophile reacts with an inaccessible leaving group to produce substitution in two-steps.



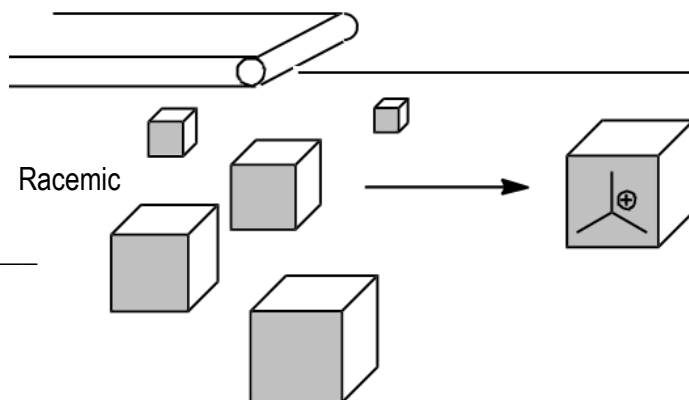
The more -R groups, the more substituted the carbocation, the more _____



S_N1 Properties (Circle One)

- Nucleophile = Strong / Weak
- Leaving Group = Unsubstituted / Highly Substituted
- Reaction coordinate = Transition State / Intermediate
- Reaction = Concerted / Two-Step
- Rate = Unimolecular / Bimolecular
- Rate = $k[\text{RX}]$ / $k[\text{Nu}][\text{RX}]$
- Stereochemistry = Inversion / Retention / Racemic
- Nickname = _____

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PRACTICE: Provide the mechanism and final products for the following reactions.

□ NOTE: Substitution reactions with *neutral nucleophiles* require an additional deprotonation step.

