

## CONCEPT: MONOSACCHARIDES — CYCLIZATION

In aqueous solutions, monosaccharides are most stable in a cyclic form.

□ Furanose = \_\_\_\_-carbon cyclic sugar

□ Pyranose = \_\_\_\_-carbon cyclic sugar

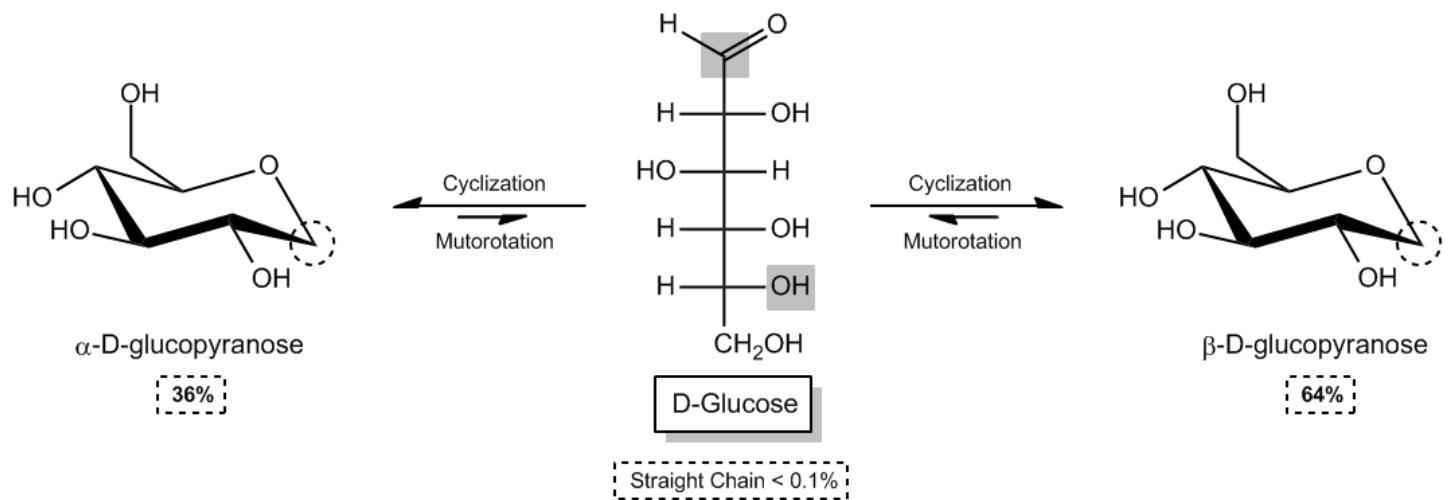
● *Nucleophilic addition of the **penultimate alcohol** to the electrophilic carbonyl carbon leads to cyclization*

□ The carbonyl carbon is \_\_\_\_\_, so it can be attacked from either the top or bottom

□ When monosaccharides *cyclize*, two different C-1 *epimers* are possible. These are known as **anomers**

- The  $\alpha$ -anomer = anomeric oxygen is \_\_\_\_\_ with the *stereodescriptor* (C-5) carbon

- The  $\beta$ -anomer = anomeric oxygen is \_\_\_\_\_ with the *stereodescriptor* (C-5) carbon



□ The OH's on the \_\_\_\_\_ in the straight chain point \_\_\_\_\_ in the ring

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- D-Sugars: *Stereodescriptor* (C-5) faces \_\_\_\_\_ on the ring

**"Downright"**

**PRACTICE:** Draw the  $\beta$ -anomer predicted through the cyclization of D-mannose.

