## **CONCEPT:** ACID-BASE CATALYSIS

• Acid catalyst increases the rate of reaction by \_\_\_\_\_ a proton to a reactant.

Acid Catalysis

Acid Catalysis  $OR \rightarrow HA \rightarrow OR \rightarrow H_2O$   $OR \rightarrow H_2O$ 

• Base catalyst increases the rate of reaction by \_\_\_\_\_ a proton from a reactant.

Base Catalysis

C-H RX R C + R-X Slow R C

- There are \_\_\_\_ types of acid-base catalysis, depending on the \_\_\_\_\_ of the catalyst.
  - 1. Specific catalysis
    - Proton transferred \_\_\_\_\_ the slow step
    - Catalyzed by \_\_\_\_\_ acid or base
    - \_\_\_\_concerted step

R LDA RX R C + R-X Slow R C enclate intermediate

## 2. General catalysis

- Proton transferred \_\_\_\_\_ the slow step
- Catalyzed by \_\_\_\_\_ acid or base
- Concerted step

## **CONCEPT:** ACID-BASE CATALYSIS

**EXAMPLE**: Draw mechanism for acid-catalyzed hydration under both specific and general acid catalysis.

PRACTICE: Provide a mechanism and appropriate bases for both, specific and general base catalysis.