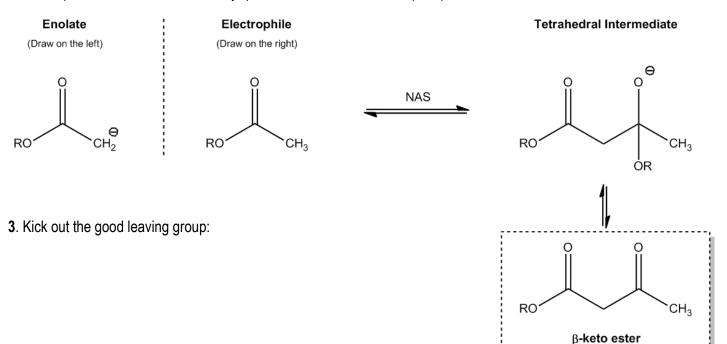
## **CONCEPT: CLAISEN CONDENSATION**

Via enolates, *esters* will react with \_\_\_\_\_\_ to condensate into \_\_\_\_\_

## Mechanism:

1. Form the enolate

2. Nucleophilic attack the other carbonyl (which we will call the electrophile)



## **Example Application:**

PRACTICE: Draw the structure of the Claisen condensation product for each of the following compounds.

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 $\underline{\mathsf{PRACTICE}}. \ \mathsf{Give} \ \mathsf{the} \ \mathsf{structure} \ \mathsf{of} \ \mathsf{the} \ \mathsf{ester} \ \mathsf{precursor} \ \mathsf{for} \ \mathsf{the} \ \mathsf{following} \ \mathsf{Claisen} \ \mathsf{condensation} \ \mathsf{product}.$