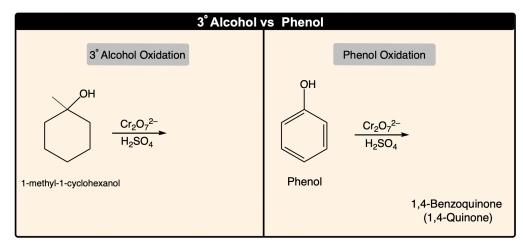
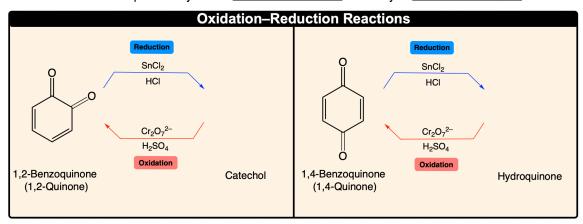
## **CONCEPT: OXIDATION OF PHENOLS TO QUINONES**

- Phenol can be oxidized into quinone in the presence of the \_\_\_\_\_ oxidizing agent \_\_\_\_\_.
  - □ Quinone is a conjugated 6-membered ring that possesses carbonyls that are \_\_\_\_\_ or \_\_\_\_ to each other.
    - Quinone is also named as \_\_\_\_\_



- Quinone can undergo reduction to yield catechol or hydroquinone.
  - □ Reduction can be accomplished by either \_\_\_\_\_ or catalytic \_\_\_\_\_



• **Note:** Both and can be oxidized into 1,4-Quinone.

**EXAMPLE:** Beginning from phenol, determine the chemical steps needed to prepare the following compound.

## **CONCEPT: OXIDATION OF PHENOLS TO QUINONES**

**PRACTICE:** Determine the products made during each of the following steps.

**PRACTICE:** Beginning from phenol, determine the chemical steps needed to prepare the following compound.