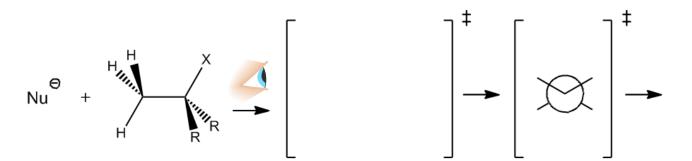
CONCEPT: E2 REACTION

□ A strong nucleophile reacts with an *inaccessible* leaving group to produce <u>beta-elimination</u> in <u>one-step</u>.



+ Nu-H

E2 Properties (Circle One)

• Nucleophile = Strong / Weak

• Leaving Group = Unsubstituted / Highly Substituted

• Reaction coordinate = Transition State / Intermediate

• Reaction = Concerted / Two-Step

• Rate = Unimolecular / Bimolecular

• Rate = k[RX] / k[Nu][RX]

• Stereochemistry =

 $\textbf{EXAMPLE:} \ \ \text{Rank the following alkyl halides in order of reactivity toward an E2 reaction}.$