

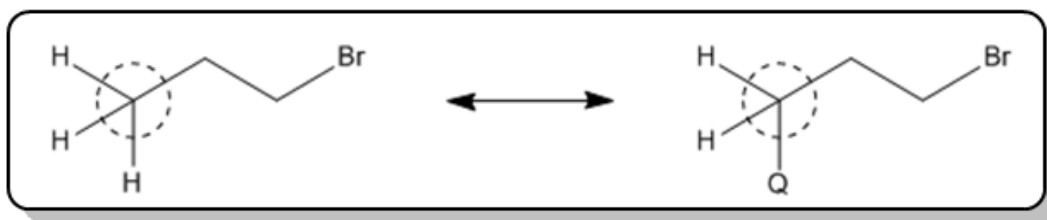
CONCEPT: ^1H NMR – PROTON RELATIONSHIPS

Hydrogens attached to the same carbon actually do have *different relationships* based on their chirality.

- The **Q-Test** is used to determine the specific type of chirality of each hydrogen.

a. Homotopic Protons

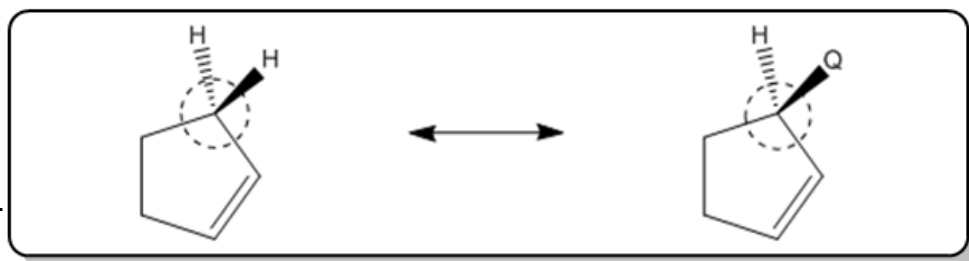
Q-Test **DOES NOT** yield new chiral center



- Protons are **always homotopic** and are considered _____ (They share a signal)
- In general, the three hydrogens on $-\text{CH}_3$ groups will always be homotopic

b. Enantiotopic Protons

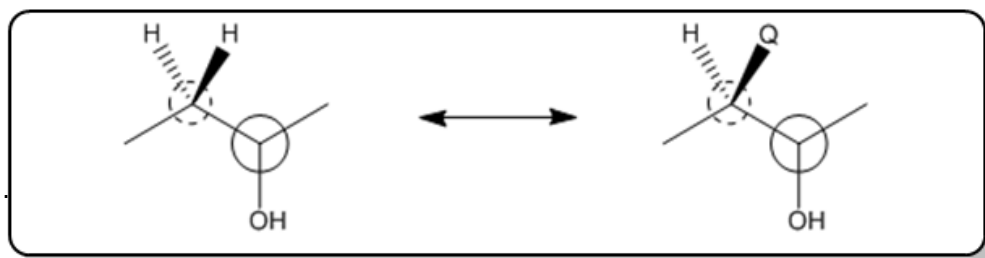
Q-Test **DOES** yield new chiral center



- No original chiral centers = protons are still _____ (They share a signal)

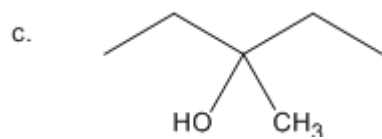
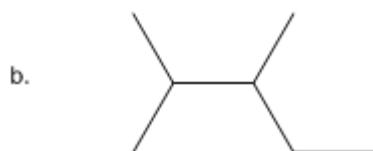
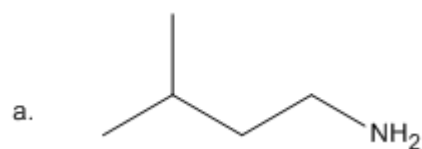
c. Diastereotopic Protons

Q-Test **DOES** yield new chiral center

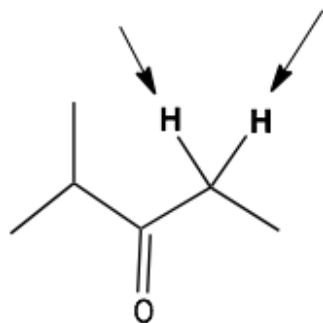


- 1+ original chiral centers = protons are now _____ (Each proton gets its own signal)

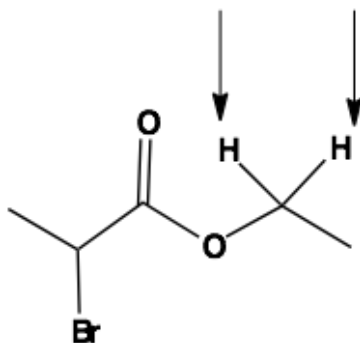
EXAMPLE: How many signals will each molecule possess in ^1H NMR?



PRACTICE: Identify the indicated set of protons as unrelated, homotopic, enantiotopic, or diastereotopic.



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