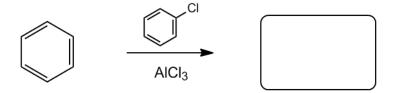
## **CONCEPT:** LIMITATIONS OF FRIEDEL-CRAFTS ALKYLATION

Friedel-Crafts Alkylation has several limitations that render it almost useless in the lab.

- 1. It does not react with *vinyl or aryl halides*. Their carbocations are far too unstable.
  - Solution: Avoid vinyl or aryl halides



- 2. Aniline derivatives ruin the Lewis Acid Catalyst
  - Solution: Avoid aniline derivatives or protect (reversibly acetylate) the amino group.

- 3. Alkylation reactions \_\_\_\_\_ the ring further reactions
  - Solution: Excess benzene or acylate instead
- 4. Alkylation reactions are susceptible to carbocation rearrangements
  - Solution: Acylate instead

## **EXAMPLE:** FC Alkylation vs. FC Acylation of benzene

<u>PRACTICE:</u> Provide the major product and the correct mechanism for the following reaction.

<u>PRACTICE:</u> Provide the major product and the correct mechanism for the following reaction.