

## CONCEPT: CALCULATING WORK IN HEAT ENGINES USING PV DIAGRAMS

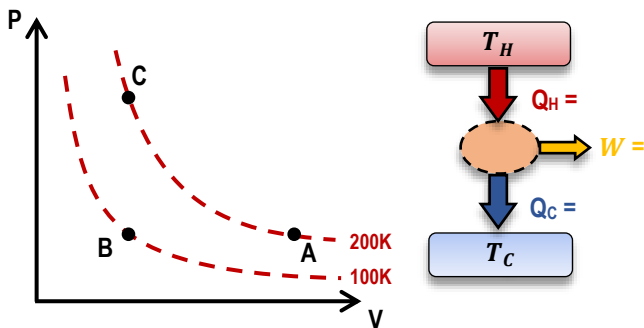
• Remember: Heat engines are cyclic processes. On PV diagrams, they always run in \_\_\_\_\_ loops.

• Remember: For heat engines,  $W = Q_H - Q_C = Q_{IN} - Q_{OUT}$ .

$Q_H$  = heat added INTO the system (from **Hot Reservoir**), i.e. the **SUM** of all [ + | - ] Qs over the cycle.

$Q_C$  = heat removed OUT of the system (to **Cold Reservoir**), i.e. the **SUM** of all [ + | - ] Qs over the cycle.

**EXAMPLE:** 2 mol of a monoatomic gas follows the cyclic process used in a heat engine. Heat is removed from the gas at constant pressure from **A**→**B**, added at constant volume from **B**→**C**, then the gas expands isothermally from **C**→**A**, doing 2300J of work, and repeats. Calculate **a)** the heat transfer of each process; **b)** the total work done by the heat engine.



Iso-P	Iso-V	Iso-T	Adiabatic (Q=0)
$\Delta E_{int}$	$Q - W$	$\Delta E_{int} = Q$	$\Delta E_{int} = -W$
$Q$	$nC_p\Delta T$	$nC_v\Delta T$	$0$
$W$	$P\Delta V$	$0$	$nRT \cdot \ln\left(\frac{V_f}{V_i}\right)$
			$-nC_v\Delta T$ OR $\frac{1}{\gamma-1}(P_iV_i - P_fV_f)$ $P_iV_i^\gamma = P_fV_f^\gamma$ $T_iV_i^{\gamma-1} = T_fV_f^{\gamma-1}$

GAS TYPE	$C_V$	$C_P$
Monoatomic	$\frac{3}{2}R$	$\frac{5}{2}R$
Diatomic	$\frac{5}{2}R$	$\frac{7}{2}R$

### EQs & Constants

$$\Delta E_{int,OF} = Q_{TO} - W_{BY}$$

$$\Delta E_{int} = 0$$

$$|W| = |Q_H| - |Q_C|$$

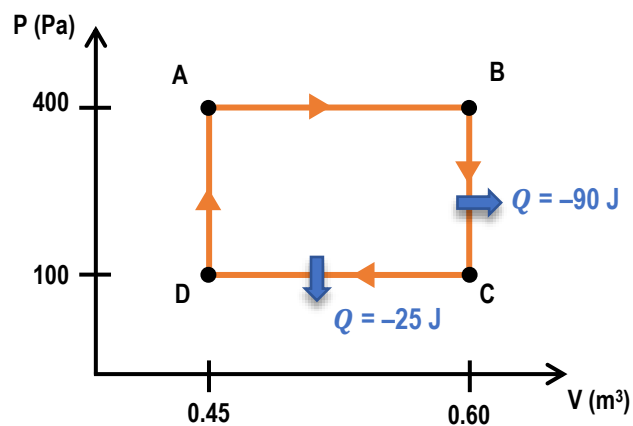
$$e = \frac{\text{output}}{\text{input}} = \frac{W}{Q_H} = 1 - \frac{Q_C}{Q_H}$$

$$K = \frac{Q_C}{W}$$

$$K_{HP} = \frac{Q_H}{W}$$

$$R = 8.314$$

**PROBLEM:** For the heat engine shown in the PV diagram below, calculate: **a)** the work done; **b)** the total heat transferred from the hot reservoir.



### HEAT ENGINES

$$\Delta E_{int} = 0$$

$$|W| = |Q_H| - |Q_C|$$

$$e = \frac{W}{Q_H} = 1 - \frac{Q_C}{Q_H}$$

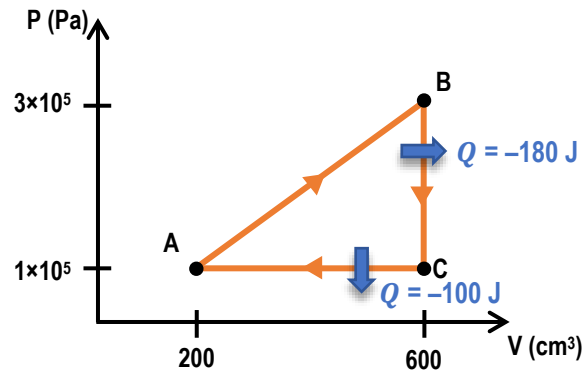
$$Q = nC_V\Delta T \quad (\text{Constant } V)$$

$$Q = nC_P\Delta T \quad (\text{Constant } P)$$

$$W = -nRT \ln\left(\frac{V_f}{V_i}\right) \quad (\text{Constant } T)$$

**PROBLEM:** A heat engine follows the cycle shown below. What is the thermal efficiency (%) of this engine?

- A) 58.8%
- B) 99.9%
- C) 14.3%
- D) 12.5%



#### HEAT ENGINES

$$\Delta E_{int} = 0$$

$$|W| = |Q_H| - |Q_C|$$

$$e = \frac{W}{Q_H} = 1 - \frac{Q_C}{Q_H}$$

$$Q = nC_V\Delta T \quad (\text{Constant } V)$$

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